May-Jun 2005

# A New Entry to Pyrazolo[4,3-*e*][1,4]diazepines. Facile Synthesis of Pyrazolo [4,3-*e*][1,4]diazepin-5,8-diones, 5,6,8-Triones and Pyrazolo[4,3-*e*]pyrrolo-[1,2-*a*][1,4]diazepin-5,10-diones

Nalla Ram Reddy, Ghanta Mahesh Reddy and Padi Pratap Reddy\*†

Department of Chemistry, Osmania University, Hyderabad-500 007, India Fax: +91 40 23095438; E-mail: reddyppou@yahoo.co.in

<sup>†</sup> Present Address: Research and Development, Dr. Reddys Laboratories Ltd., API, Unit-IV, Plot No. 9/A,
Phase-III, IDA, Jeedimetla, Hyderabad-500 055, India
Received July 1, 2004

A facile approach to pyrazolo[4,3-e][1,4]diazepin-5,8-diones and pyrazolo[4,3-e]pyrrolo[1,2-a][1,4]-diazepin-5,10-diones is reported. Strategy involved the utility of  $\alpha$ -amino acid as a three-atom segment in the construction of diazepine skeleton on the preformed pyrazole ring.

J. Heterocyclic Chem., 42, 675 (2005).

## Introduction.

Immense pharmacological importance of [1,4]benzodiazepines [1] led to a great deal of work on various facets of heteroannelated[1,4]diazepines. Thus, [1,4]diazepines fused to thiophenes [2], imidazoles [3], pyrroles [4], isoxazoles [5] and pyrazoles [6-7] were synthesized and investigated for their pharmacological activity.

Our current interest in fused pyrazoles coupled with above findings prompted us to plan the synthesis of pyrazolo[4,3-e][1,4]diazepine derivatives. Earlier, Dewald and co-workers [6] reported the synthesis of 1,3/2,3-dialkyl-4,6-dihydro-8-aryl pyrazolo[4,3-e][1,4]diazepine-5-ones by making use of 1,3/2,3-dialkyl-4-aminopyrazolyl aryl ketones as intermediates. In the present paper, we describe the synthesis of various new fused pyrazolodiazepine derivatives through a short synthetic sequence. Utility of  $\alpha$ -amino acids as a three atom segment in the construction of diazepine skeleton on the preformed pyrazole ring served as a facile route to pyrazolo[4,3-e][1,4]diazepines. 1-Alkyl-4-nitro-3-n-propyl pyrazolyl-5-carboxylic acid [8] **1a/1b** was chosen as the precursor.

## Results and Discussion.

Reaction of **1a** with thionyl chloride and subsequent condensation of the resulting acid chloride with glycine in 10% KOH solution afforded 1-methyl-3-*n*-propyl-1*H*-pyrazolyl-5-carboxamido)acetic acid **2a**. Treatment of **2a** with hydrogen in the presence of Raney nickel at 75 psi pressure directly furnished 1-methyl-3-*n*-propyl-1,4,5,6,7,8-hexahydropyrazolo[4,3-*e*][1,4]diazepin-5,8-dione **3a** in 73% yield. The structural assignment of **3a** was based on its ir, mass, <sup>1</sup>H and <sup>13</sup>C nmr spectral data.

In the mass spectrum of 3a the highest peak at m/e 222 corresponds to the molecular ion. Presence of two distinct amide functions in the compound is deduced from the ir spectrum [KBr, 3178, 3070 cm<sup>-1</sup> (two amide NH); 1689, 1669 cm<sup>-1</sup> (amide carbonyls)]. <sup>1</sup>H nmr spectrum (DMSO-d<sub>6</sub>) displayed two amide proton signals at  $\delta$  10.1 (br s, 1H)

and  $\delta$  8.3 (t, 1H) while doublet at  $\delta$  3.6 integrating for two protons is assignable to glycine CH<sub>2</sub> group. Other signals in the <sup>1</sup>H nmr spectrum are due to *N*-methyl ( $\delta$  4.0, s, 3H) and *n*-propyl [ $\delta$  1.0 (t, 3H, CH<sub>3</sub>), 1.6 (m, 2H, CH<sub>2</sub>), 2.6 (t, 2H, CH<sub>2</sub>)]. <sup>13</sup>C nmr spectrum of **3a** showed a total of ten signals. They include signals in the downfield region due to two carbonyl carbons ( $\delta$  168.6 and 161.7) and three carbons of pyrazole nucleus ( $\delta$  148.6, 125.1 and 121.6). In the upfield region, the five signals at  $\delta$  13.3, 21.5, 26.4, 38.2 and 45.3 are due to *n*-propyl, *N*-methyl and *N*-methylene carbons. Thus, in the Raney nickel reduction of nitropyrazolylglycine derivative **2a** at 75 psi pressure, initially formed aminopyrazolylglycine intermediate **4** is undergoing concomitant dehydrocylisation under the reaction conditions to provide the pyrazolodiazepine dione **3a** in one

step. To verify this, compound **2a** was subjected to Raney Ni reduction under mild conditions (at atmospheric pressure) and open chain intermediate **4** was isolated as the product. Compound **4** smoothly cyclised into **3a** in the presence of 1,3-dicyclohexylcarbodiimide.

Three other α-aminoacids, L-alanine, L-valine and L-phenylalanine furnished the corresponding pyrazolodiazepines **3b-d** in good yields (Scheme-1). 1-Ethyl pyrazole derivative **1b** provided the corresponding pyrazolodiazepine derivative **3e** by participating in reaction with glycine through a similar synthetic sequence. A similar two step synthetic operation starting from **1a/1b** and L-proline yielded novel fused tricyclic compounds, pyrazolo[4,3-*e*]-pyrrolo[1,2-*a*]diazepin-5,10-diones **3f-g** (Scheme-2).

Condensation of 1-alkyl-4-amino-3-*n*-propylpyrazole-5-carboxamide **5a/5b** with oxalyl chloride in dichloromethane in the presence of pyridine directly yielded 1-alkyl-3-*n*-propyl-1,4,5,6,7,8-hexahydropyrazolo[4,3-*e*]-[1,4]diazepin-5,6,8-triones **6a/6b** (Scheme-3). All the pyrazolodiazepines were characterized based on their ir, mass and <sup>1</sup>H nmr spectral data (Table-1).

## Conclusion.

Thus, we have provided a facile new entry to pyrazolo[4,3-*e*][1,4]diazepines. Novel fused tricyclic compounds, pyrazolo[4,3-*e*]pyrrolo[1,2-*a*]diazepines were obtained by this synthetic procedure.

### Scheme 2

Table 1

Physical and Spectral Data of 2-(1-Alkyl-4-nitro-3-*n*-propyl-1*H*-pyrazolyl-5-carboxamido)substituted
Carboxylic Acids **2**, Pyrazolodiazepindiones **3** and Pyrazolodiazepintriones **6** 

Compd. No.	m.p. (°C)	Yield (%)	ir	¹H nmr	MS (m/z) (M <sup>+</sup> )
2a	174	85	3246 1725 1654	1.0 (t, 3H, CH <sub>3</sub> ), 1.7 (m, 2H, CH <sub>2</sub> ), 2.85 (t, 2H, CH <sub>2</sub> ), 3.9 (s, 3H, N-CH <sub>3</sub> ), 4.1 (d, 2H, N-CH <sub>2</sub> ), 8.9 (br s, 1H, NH)	270
2b	139	83	3306 1709 1650	$\begin{array}{c} 1.0 \;\; (t,\; 3H,\; CH_3),\; 1.6 \;\; (m,\; 2H,\; CH_2),\; 2.4 \;\; (d,\; 3H,\; CH_3),\; 2.7 \;\; (t,\; 2H,\; CH_2),\; 4.0 \;\; (s,\; 3H,\; N\text{-}CH_3),\; 4.4 \;\; (m,\; 1H,\; CH),\; 8.5 \;\; (br\; s,\; 1H,\; NH) \end{array}$	284
<b>2</b> c	78	76	3271 1725 1646	1.0 (m, 9H, 3 x CH <sub>3</sub> ), 1.8 (m, 3H, CH and CH <sub>2</sub> ), 2.85 (t, 2H, CH <sub>2</sub> ), 3.95 (s, 3H, N-CH <sub>3</sub> ), 4.7 (m, 1H, CH), 8.4 (d, 1H, NH)	312
2d	183	71	3268 1703 1654	1.0 (t, 3H, CH <sub>3</sub> ), 1.7 (m, 2H, CH <sub>2</sub> ), 2.8 (t, 2H, CH <sub>2</sub> ), 3.2 (m, 2H, Ar-CH <sub>2</sub> ), 4.0 (s, 3H, N-CH <sub>3</sub> ), 4.2 (m, 1H, CH), 7.2 (m, 5H, Ar-H), 8.0 (br d, 1H, NH)	360
2e	160	81	3247 1721 1662	0.95 (t, 3H, CH <sub>3</sub> ), 1.35 (t, 3H, N-C-CH <sub>3</sub> ), 1.6 (m, 2H, CH <sub>2</sub> ), 2.4 (t, 2H, CH <sub>2</sub> ), 4.1 (d, 2H, N-CH <sub>2</sub> ), 4.3 (q, 2H, N-CH <sub>2</sub> ), 8.6 (br s, 1H, NH)	284
2f	55	71	1734 1654	1.0 (t, 3H, CH <sub>3</sub> ), 1.8 (m, 2H, CH <sub>2</sub> ), 2.2-2.6 (m, 4H, CH <sub>2</sub> and two pyrrolidine protons), 2.9 (m, 2H, CH <sub>2</sub> ), 3.4 (br s, 2H), 3.9 (s, 3H, N-CH <sub>3</sub> ), 4.8 (t, 1H, C <sub>5</sub> -H), 8.9 (br s, 1H, OH)	310
2g	48	76	1725 1651	1.0 (t, 3H, CH <sub>3</sub> ), 1.45 (t, 3H, N-C-CH <sub>3</sub> ), 1.7 (m, 2H, CH <sub>2</sub> ), 2.1 (m, 2H, CH <sub>2</sub> ), 2.2 (m, 1H), 2.4 (m, 3H, CH and CH <sub>2</sub> ), 2.8 (m, 2H), 4.2 (m, 2H, CH <sub>2</sub> ), 4.8 (m, 1H, NH), 9.8 (br s, 1H, OH)	324

Table 1 (continued)

Compd. No.	m.p. (°C)	Yiel d (%)	ir	¹H nmr	MS (m/z) (M <sup>+</sup> )	CHNAnalysis
3a	199 [a]	81	3178 3070 1689 1669	$\begin{array}{c} 1.0 \ (t, 3H, CH_3), 1.6 \ (m, 2H, CH_2), 2.6 \ (t, 2H, CH_2),\\ 4.0 \ (s, 3H, N\text{-}CH_3), 3.6 \ (d, 2H, N\text{-}CH_2), 8.3 \ (br \ s, 1H, NH), 10.1 \ (br \ s, 1H, NH) \end{array}$	222	Calcd: C, 54.04; H, 6.35; N, 25.21. Found: C, 54. 19; H, 6.33; N, 25.29.
3b	215 [a]	78	3326 3155 1699 1669	1.0 (t, 3H, CH <sub>3</sub> ), 1.7 (m, 2H, CH <sub>2</sub> ), 2.4 (d, 3H, CH <sub>3</sub> ), 2.7 (t, 2H, CH <sub>2</sub> ), 4.0 (s, 3H, NCH <sub>3</sub> ), 4.6 (m, 1H, CH), 7.4 (br s, 1H, NH), 7.9 (br s, 1H, NH)	236	Calcd: C, 54.92; H, 6.83; N, 23.71. Found: C, 55.11; H, 6.82; N, 23.78.
3c	159 [a]	75	3271 3108 1703 1696	1.1 (m, 9H, 3 x CH <sub>3</sub> ), 1.7 (m, 2H, CH <sub>2</sub> ), 2.3 (m, 1H, CH), 2.9 (t, 2H, CH <sub>2</sub> ), 4.0 (s, 3H, N-CH <sub>3</sub> ), 4.6 (d, 1H, CH), 8.0 (d, 1H, NH), 8.3 (br s, 1H, NH)	264	Calcd: C, 59.07; H, 7.63; N, 21.20. Found: C, 59.24; H, 7.61; N, 21.27.
3d	89 [a]	69	3281 3095 1697 1649	0.9 (t, 3H, CH <sub>3</sub> ), 1.7 (m, 2H, CH <sub>2</sub> ), 2.6 (t, 2H, CH <sub>2</sub> ), 3.1 (d, 2H, Ar-CH <sub>2</sub> ), 3.6 (m, 1H, CH), 4.0 (s, 3H, N-CH <sub>3</sub> ), 7.3 (m, 5H, Ar-H), 8.6 (br s, 1H, NH), 10.2 (br s, 1H, NH)	312	Calcd: C, 65.37; H, 6.45; N, 17.94. Found: C, 65.53; H, 6.44; N, 17.99.
3e	163 [a]	77	3324 3153 1694 1666	1.0 (t, 3H, CH <sub>3</sub> ), 1.4 (t, 3H, N-C-CH <sub>3</sub> ), 1.7 (m, 2H, CH <sub>2</sub> ), 2.6 (t, 2H, CH <sub>2</sub> ), 3.8 (d, 2H, N-CH <sub>2</sub> ), 4.4 (q, 2H, N-CH <sub>2</sub> ), 7.5 (br, t, 1H, NH), 9.3 (br s, 1H, NH)	236	Calcd: C, 55.92; H, 6.83; N, 23.71. Found: C, 56.09; H, 6.83; N, 23.77.
3f	183 [a]	80	3160 1684 1653	1.0 (t, 3H, CH <sub>3</sub> ), 1.6 (m, 2H, CH <sub>2</sub> ), 2.6 (m, 2H, CH <sub>2</sub> ), [2.0 (br s, 3H), 2.8 (m, 1H), 3.6 (m, 2H), 4.1 (m, 1H, due to pyrrolidine protons)], 4.0 (s, 3H, N-CH <sub>3</sub> ), 9.4 (br s, 1H, NH)	262	Calcd: C, 59.53; H, 6.92; N, 21.36. Found: C, 59.68; H, 6.90; N, 21.43.
<b>3</b> g	164 [a]	83	3169 1683 1644	1.0 (t, 3H, CH <sub>3</sub> ), 1.4 (t, 3H, N-C-CH <sub>3</sub> ), 1.7 (m, 2H, CH <sub>2</sub> ), 2.7 (t, 2H, CH <sub>2</sub> ), [2.0 (br s, 3H), 2.8 (br s, 1H), 3.6 (br s, 2H), 4.1 (br d, 1H pyrrolidine protons)], 4.4 (q, 2H, N-CH <sub>2</sub> ), 9.8 (br s, 1H, NH)	276	Calcd: C, 60.85; H, 7.30; N, 20.28. Found: C, 60.96; H, 7.31; N, 20.32.
6a	297 [b]	84	3292 1698 1672 1691	$1.0$ (t, $3H$ , $CH_3$ ), $1.7$ (m, $2H$ , $CH_3$ ), $2.5$ (t, $2H$ , $CH_2$ ), $4.0$ (s, $3H$ , $N$ - $CH_3$ ), $7.6$ (br s, $1H$ , $NH$ ), $10.5$ (br s, $1H$ , $NH$ )	236	Calcd: C, 50.84; H, 5.12; N, 23.72. Found: C, 50.93, H, 5.11; N, 23.79.
6b	<b>265</b> [c]	80	3253 1691 1662 1623	1.0 (t, 3H, CH <sub>3</sub> ), 1.4 (t, 3H, N-C-CH <sub>3</sub> ), 1.6 (m, 2H, CH <sub>2</sub> ), 2.4 (t, 2H, CH <sub>2</sub> ), 4.3 (m, 2H, N-CH <sub>2</sub> ), 7.5 (br s, 1H, NH), 10.1 (br s, 1H, NH)	250	Calcd: C, 52.79; H, 5.64; N, 22.39. Found: C, 52.90; H, 5.62; N, 22.44.

Solvent used for recrystallization: [a] ethyl acetate:n-hexane; [b] methanol:ethyl acetate; [c] methanol.

## Scheme 3

$$H_2N$$
 $H_2N$ 
 $H_2N$ 
 $CH_2CH_2CH_3$ 
 $CH_2CH_2CH_3$ 
 $CH_2CH_3$ 
 $CH_3$ 
 $C$ 

# **EXPERIMENTAL**

 $^{1}\mathrm{H}$  and  $^{13}\mathrm{C}$  nmr spectra were recorded on a Varian Gemini (300 and 100 MHz respectively) nmr spectrometer at ambient temperature using TMS as internal standard. Mass spectrometry (70 eV) was carried out on a Perkin-Elmer Hitachi RMU-6 L

instrument. IR spectra were obtained in KBr pellets on a Shimadzu 435 instrument. Elemental Analysis was carried out on Perkin-Elmer 2400 S CHN analyzer. Melting points were determined in capillaries using Polman digital melting point apparatus (Model-mp-96). Reagents and solvents were of analytical grade. Solvents were dried before use.

2-(1-Alkyl-4-nitro-3-*n*-propyl-1*H*-pyrazolyl-5-carboxamido)-substituted Carboxylic Acids (**2a-g**).

## General Procedure.

A mixture of 1-alkyl-4-nitro-3-n-propyl pyrazolyl-5-carboxylic acid 1a/1b [8] (0.01 mol) and thionyl chloride (10 mL) in benzene (10 mL) was refluxed for 3 h. The reaction mixture was cooled and excess thionyl chloride was removed *in vacuo*. The oily residue was dissolved in benzene (10 mL) and this acid chloride was cautiously added to the appropriate  $\alpha$ -amino acid (0.011 mol) in 10% aq. KOH solution (20 mL) at 10-20 °C and reaction mixture was stirred for 15 min. The organic layer was separated

and aqueous layer was acidified with 2 N HCl (10 mL) and extracted with ethyl acetate (2 x 25 mL). Ethyl acetate solution was dried (Na<sub>2</sub>SO<sub>4</sub>), concentrated and the resulting carboxylic acid derivative 2 was recrystallised from ethyl acetate/hexane.

1-Alkyl-6-substituted-3-*n*-propyl-1,4,5,6,7,8-hexahydropyrazolo-[4,3-*e*][1,4]diazepin-5,8-diones (**3a-g**).

#### General Procedure.

To a solution of **2** (5 mmol) in methanol (25 mL) was added Raney nickel (0.5 g) and reaction mixture was placed under a hydrogen atmosphere (75 psi) in a Paar hydrogenation apparatus for 4-5 h, and then filtered through a celite bed. The catalyst was washed with methanol (15 mL) and combined filtrates were evaporated to dryness *in vacuo*. The resulting residue was taken in  $\rm H_2O$  (20 mL) and extracted with ethyl acetate (3 x 25 mL). The organic extracts were combined, dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated under reduced pressure to give the corresponding pyrazolodiazepine dione derivative **3a-g** which was recrystallised from appropriate solvent.

2-(4-Amino-1-methyl-3-*n*-propyl-1*H*-pyrazolyl-5-carboxamido) Acetic Acid (**4**).

To a solution of 2-(1-methyl-4-nitro-3-n-propyl-1H-pyrazolyl-5-carboxamido) acetic acid 2a (2.7 g, 5 mmol) in methanol (25 mL), was added Raney nickel (0.5 g) and reaction mixture was placed under a hydrogen atmosphere at atmospheric pressure for 4 h and then filtered through a celite bed. The catalyst was washed with methanol (10 mL) and the combined filtrates were evaporated to dryness  $in\ vacuo$ . The resulting residue was taken in water (25 mL) and extracted with chloroform (3 x 25 mL). The organic extracts were combined, dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated under reduced pressure. Recrystallisation of the residue from benzene gave compound 4, yield 72%, m.p.124 °C. MS: m/e 240 M+; IR (KBr, cm-1) 3510, 3366, 3234, 1727, 1646;  $^1$ H NMR (CDCl<sub>3</sub>):  $\delta$  1.0 (t, 3H, CH<sub>3</sub>), 1.65 (m, 2H, CH<sub>2</sub>), 2.6 (t, 2H, CH<sub>2</sub>), 3.8-4.2 (br m, 7H, N-CH<sub>3</sub>, N-CH<sub>2</sub>, NH<sub>2</sub>), 8.8 (s, 1H, NH).

Cyclisation of 2-(4-Amino-1-methyl-3-*n*-propyl-1*H*-pyrazolyl-5-carboxamido) Acetic Acid (4).

A solution of 4 (0.01 mol) and catalytic amount of 4-dimethylamino pyridine in dichloromethane (20 mL) was cooled to 0 °C.

Solution of DCC (2.06 g, 0.01 mol) in dichloromethane was added drop wise to the above solution over a period of 10 min while maintaining the temperature at 0-5 °C. After the addition, the solution was stirred for an additional period of 5 min and the temperature was allowed to rise to room temperature over a period of 1 h. Separated urea derivative was filtered, washed with dichloromethane (15 mL), combined organic layers were washed with water, dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated *in vacuo*. The residue was triturated with *n*-hexane and the resulting solid was recrystallised from ethyl acetate to give pyrazolo[4,3-*e*][1,4]diazepin-5,8-dione derivative **3a**, yield 68%, m.p. 199 °C.

1-Methyl/ethyl-3-*n*-propyl-1,4,5,6,7,8-hexahydropyrazolo[4,3-*e*]-[1,4]diazepin-5,6,8-trione (**6a/b**).

## General Method.

To a solution of 4-amino-1-alkyl-3-*n*-propyl pyrazolyl-5-carboxamide **5a/b** (0.01 mol) in dichloromethane (40 mL) and pyridine (2 mL), oxalyl chloride (1.3 g, 0.011 mol) was added drop wise and reaction mixture was stirred at room temperature for 3 h. 1-Alkyl-3-*n*-propyl-1,4,5,6,7,8-hexahydropyrazolo[4,3-*e*][1,4]-diazepin-5,6,8-trione 6a/b separated out from the clear solution, which was collected by filtration, washed with water (2 x 30 mL) and recrystallised from suitable solvent.

## REFERENCES AND NOTES

- \* To whom correspondence should be addressed.
- [1] J. A. Vida, Medicinal Chemistry, Part III, M. W. Wolf, I. W. Burger, John

Wiley & Sons, New York 1981, pp. 787.

- [2] F. J. Tinney, J. P. Sanchez and J. A. Noas, J. Med. Chem., 17, 624 (1974).
  - [3] A. Edenhofer, Helv. Chem. Acta., 58, 2192 (1975).
- [4] L. Fontanella, L. Mariani, G. Tarzia and N. Corsico, *Eur. J. Med. Chem.*, **11**, 217 (1976).
  - [5] R. Jaunin, Helv. Chem. Acta., 57, 1934 (1974).
- [6] H. A. DeWald, I. C. Nordin, Y. J. L' Italien and R. F. Parcel, *J. Med. Chem.*, **16**, 1346 (1973).
- [7] H. A. DeWald , S. Lobbestael and B. P. H. Poschel, *J. Med. Chem.*, **24**, 982, (1981).
- [8] J. Fray, D. J. Bull, K. Cooper, M. J. Parry and M. H. Stefaniak, *J. Med. Chem.*, **38**, 3524 (1995).